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remain, even though chemical activity be absent, so also the peculiar molecular constitution, characteristic of life may remain even though life—*i. e.*, active life—be absent. As chemism may be completely destroyed by cold and revived by heat, if the necessary condition, *viz.*, a certain equivalent composition, remains: so life may be completely destroyed by cold, and again revived by heat if so be the necessary condition—a certain molecular constitution remains.

Again, Mr. Brown, in the article referred to, says that Spencer's definition of life, *viz.*, 'a continuous adjustment of internal to external conditions' must be revised, since it applies only to *active* life and not to suspended life. I think not. Life in the true sense, *i. e.*, *actual* life must be active—the essential nature of life, as of all energy, is *activity*; but there is a necessary underlying condition, *i. e.*, a peculiar molecular constitution, which may be called *potential life*. As equivalent composition is potential chemism, which may under certain conditions become actual chemism, so a peculiar molecular constitution of protoplasm is potential life, which may under certain conditions become actual life. *Death* is not merely an *extinction* of life, *i. e.*, actual life, but also the *destruction* of the necessary condition of life, *i. e.*, the characteristic molecular structure of protoplasm. Extinction of life, therefore, is not necessarily *death*. There are therefore three conditions of protoplasm, *viz.*: (1) *living*, a condition in which life is actual; (2) *potentially living*, in which the necessary molecular constitution or vital constitution is present; and (3) *dead*, in which the vital constitution also is wanting.\*

JOSEPH LE CONTE.

BERKELEY, CAL., May 24, 1901.

#### A 'SAND-BOW'—AN UNUSUAL OPTICAL PHENOMENON.

THE following description, based on personal observation, is presented without discussion of the optical principles involved.

\* Similar views on the molecular constitution of living protoplasms were brought out by me, in January, 1892 in a lecture before the Philosophical Society of Washington. *Bulletin*, Vol. XII., p. 29, 1892.

On the evening of May 16, the writer was crossing the main ridge of Antelope Island—the largest land body within the area of the Great Salt Lake. As he began the descent on the eastern slope, there appeared between the island and the mainland what seemed at first glance to be segment of a brilliant rainbow of unusual width. It was evident, however, that no rain was falling in that direction. Clouds were gathering in the south and west, but the sun was yet unobscured. A wind setting toward the mainland had lifted from the dry flats large quantities of the 'oolitic sand,' with which the lake bottom and the recently dried patches on this side of the island are covered to a depth varying from a few inches to several feet. This so-called 'sand' consists of calcareous spherules, fairly uniform in size between the limits of No. 8 and No. 10 shot. The oolitic bodies are polished and exhibit a pearly luster.

It would seem that the outer spherical surfaces reflected the light in such a manner as to produce the bow. The colored column appeared almost to touch the lake bed, and its ends subtended with the observer an angle of about 40°. The prismatic colors were distinct, the red being outside, *i. e.*, away from the sun. In apparent width the column was fully double that of the ordinary rainbow. A fainter secondary bow was plainly visible beyond the primary, with the colors in reverse order. The phenomenon was so brilliant as to attract the attention of all members of the party, and it remained visible for over five minutes, then, as the sun sank lower, it rapidly died away.

The production of a color bow by reflection from the outer surfaces of opaque spherules is a new phenomenon to the writer. It is inexplicable on the principle of refraction and total reflection from the interior of transparent spheroids, according to which the rainbow is generally explained.

If phenomena similar or analogous to the foregoing have been observed, reports of the same would doubtless be of instructive interest.

JAMES E. TALMAGE.

UNIVERSITY OF UTAH,  
SALT LAKE CITY, May 20, 1901.